

DE LA RECHERCHE À L'INDUSTRIE



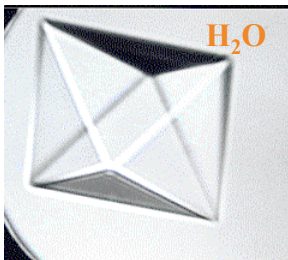
Caractériser la matière à des pressions extrêmes

*DÉCOUVERTE D'UN « CLATHRATE » DE
VAN DER WAALS : $(N_2)_6NE_7$*

Thomas Plisson, Gunnar Weck, Paul Loubeyre
CEA, DAM, DIF
Seminaire CEA – 2014

HIGH PRESSURE

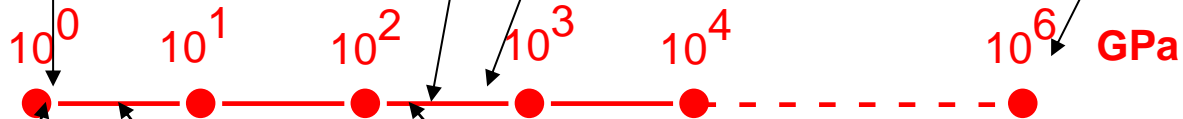
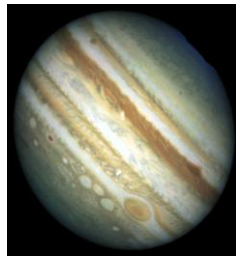
P~1.3 GPa



P~380 GPa



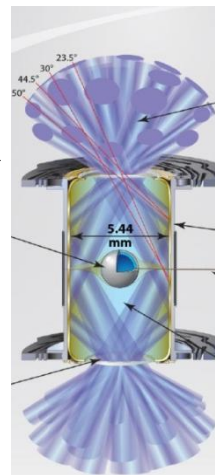
P~700 GPa



P~0.1 GPa



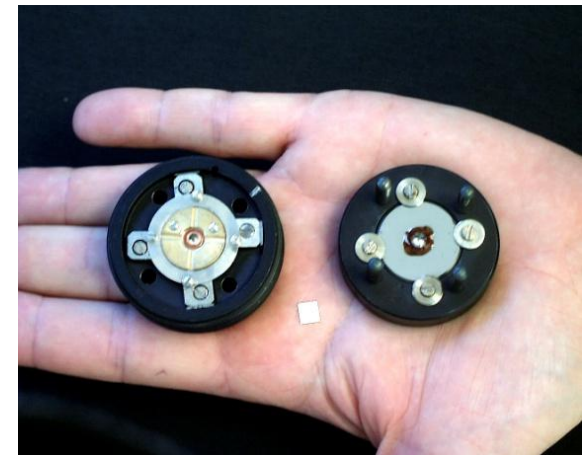
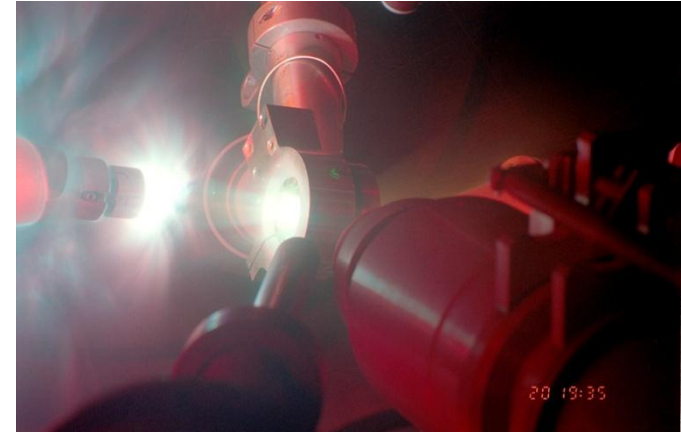
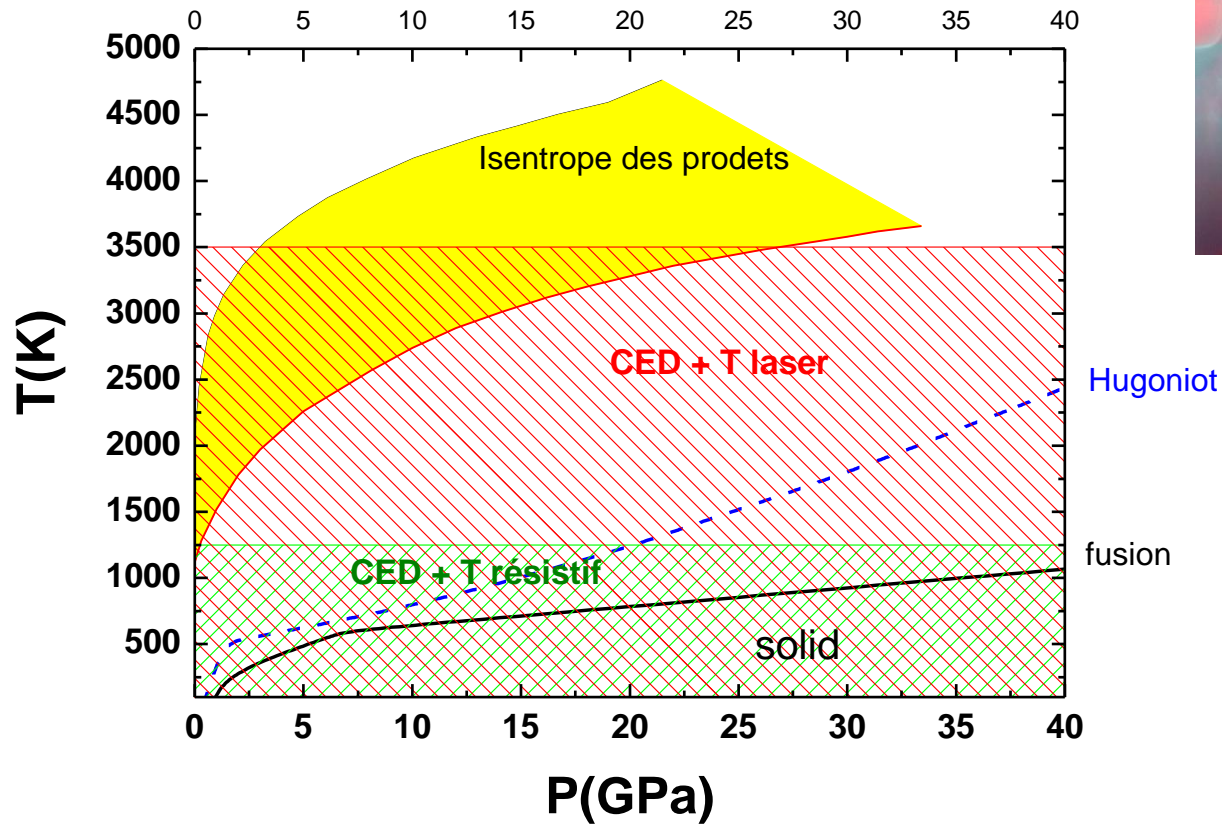
P~6GPa



P~200GPa

1 bar = 10⁵ Pa
 10 kbar = 1 GPa
 1 Mbar = 100 GPa

STATIC VS DYNAMIC



EXAMPLES OF APPLICATIONS

Hydrogen and hydrides

Are there new quantum many-body behaviors at high pressure? Metal H ?

Optical studies of solid hydrogen to 320 GPa and evidence for black hydrogen

Paul Loubeyre^a, Florent Occelli^b & René LeToullec^{c,d}

A little bit of lithium does a lot for hydrogen

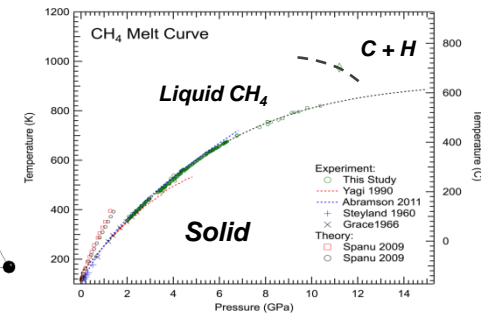
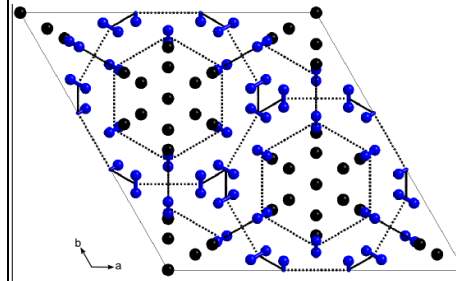
Eva Zurek^{a,1}, Roald Hoffmann^{a,2}, N. W. Ashcroft^b, Artem R. Oganov^{c,d}, and Andriy O. Lyakhov^e

^aDepartment of Chemistry and Chemical Biology, Baker Laboratory, and ¹Laboratory of Atomic and Solid State Physics and Cornell Center for Materials Research, Cornell University, Ithaca, NY 14853; ²Department of Geosciences and New York Center for Computational Science, Stony Brook University, Stony Brook, NY 11794-2100; and ³Geology Department, Moscow State University, Moscow 11992, Russia

Contributed by Roald Hoffmann, July 28, 2009 (sent for review June 29, 2009)

New materials / High pressure chemistry

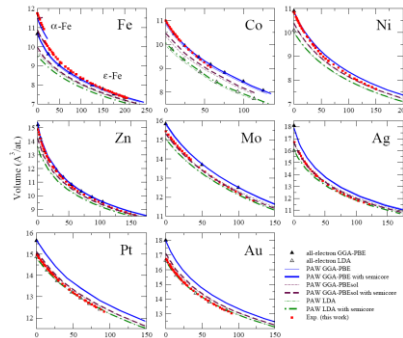
100 GPa ~ 1 eV (E of chemical bonding).



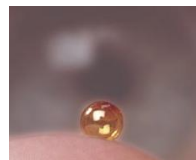
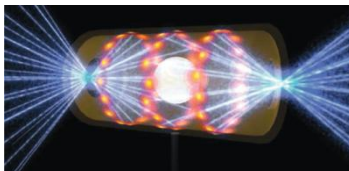
Equations of state

Are ab-initio calculations reliably predictive?

➤ Metals



➤ C-H



Fluids

PHYSICAL REVIEW B, VOLUME 65, 174105

Quantitative structure factor and density measurements of high-pressure fluids in diamond anvil cells by x-ray diffraction: Argon and water

Jon H. Eggert,^{1,2} Gunnar Weck,¹ Paul Loubeyre,¹ and Mohamed Mezouar³
¹DIF/DPTA/SPMC, CEA, 91680 Bruyères-le-Châtel, France

NATURE | VOL 409 | 15 JANUARY 2006 | www.nature.com

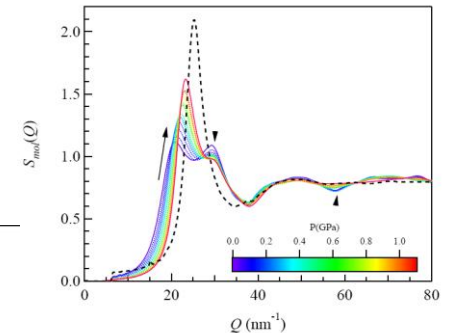
A first-order liquid-liquid phase transition in phosphorus

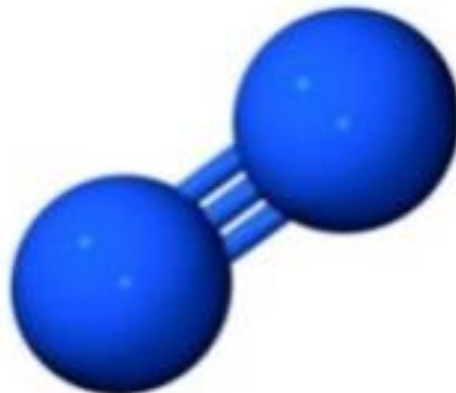
Yoshinori Katayama¹, Takeshi Mizutani¹, Wataru Utsumi¹, Osamu Shimomura¹, Masaki Yamakita¹ & Ken-ichi Funakoshi¹

26 APRIL 2013 VOL 340 SCIENCE

Melting of Iron at Earth's Inner Core Boundary Based on Fast X-ray Diffraction

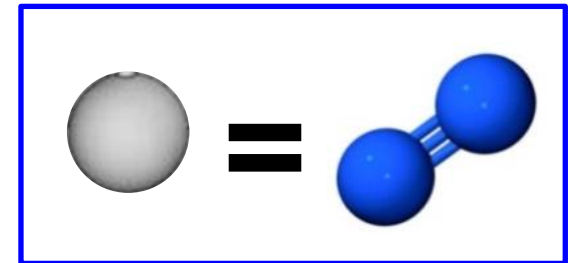
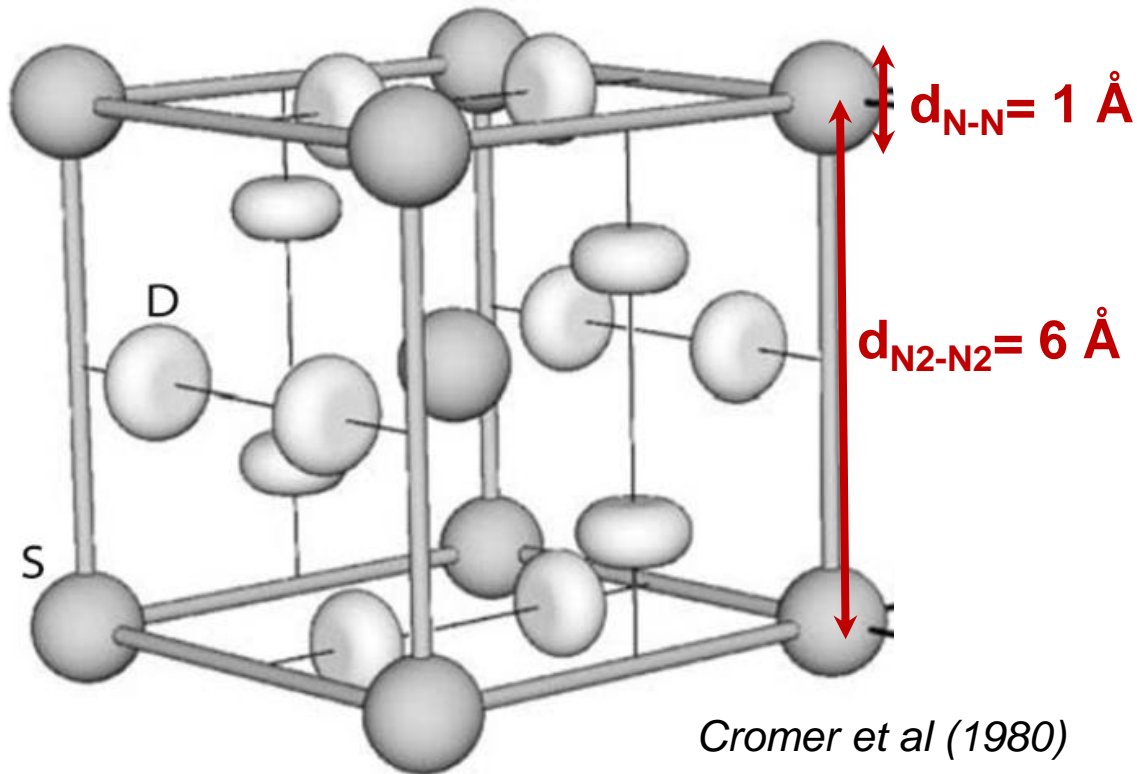
S. Anzellini,^{1*} A. Dewaele,¹ M. Mezouar,² P. Loubeyre,¹ G. Morard³





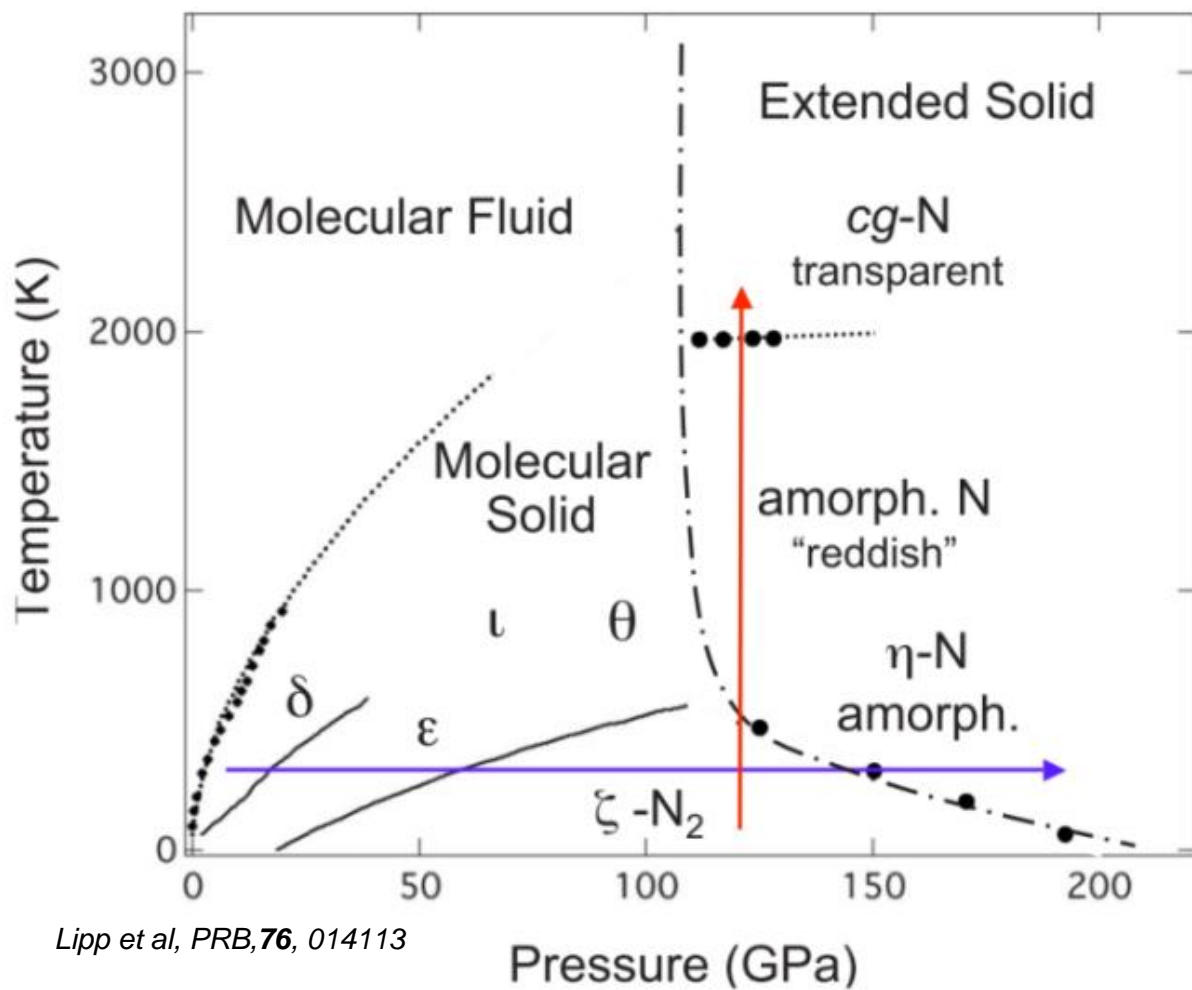
N_2 triple bond : 9.8 eV/atom

Structure of solid nitrogen at 4.9 GPa / 300 K = cubic ($Pm-3n$)



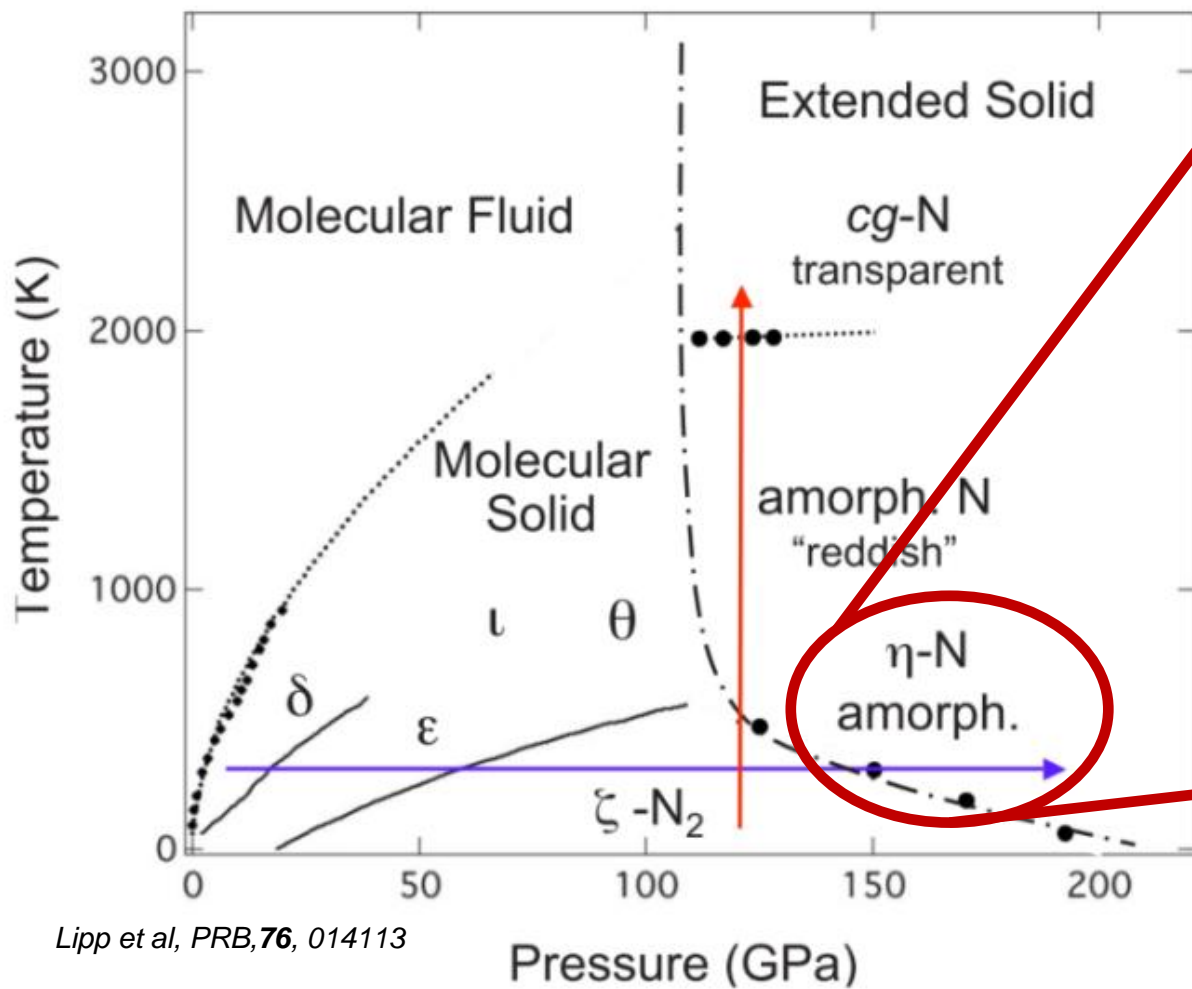
«van der Waals solid» = Molecules bonded by (weak) van der Waals forces

Evolution under pressure



Lipp et al, PRB, 76, 014113

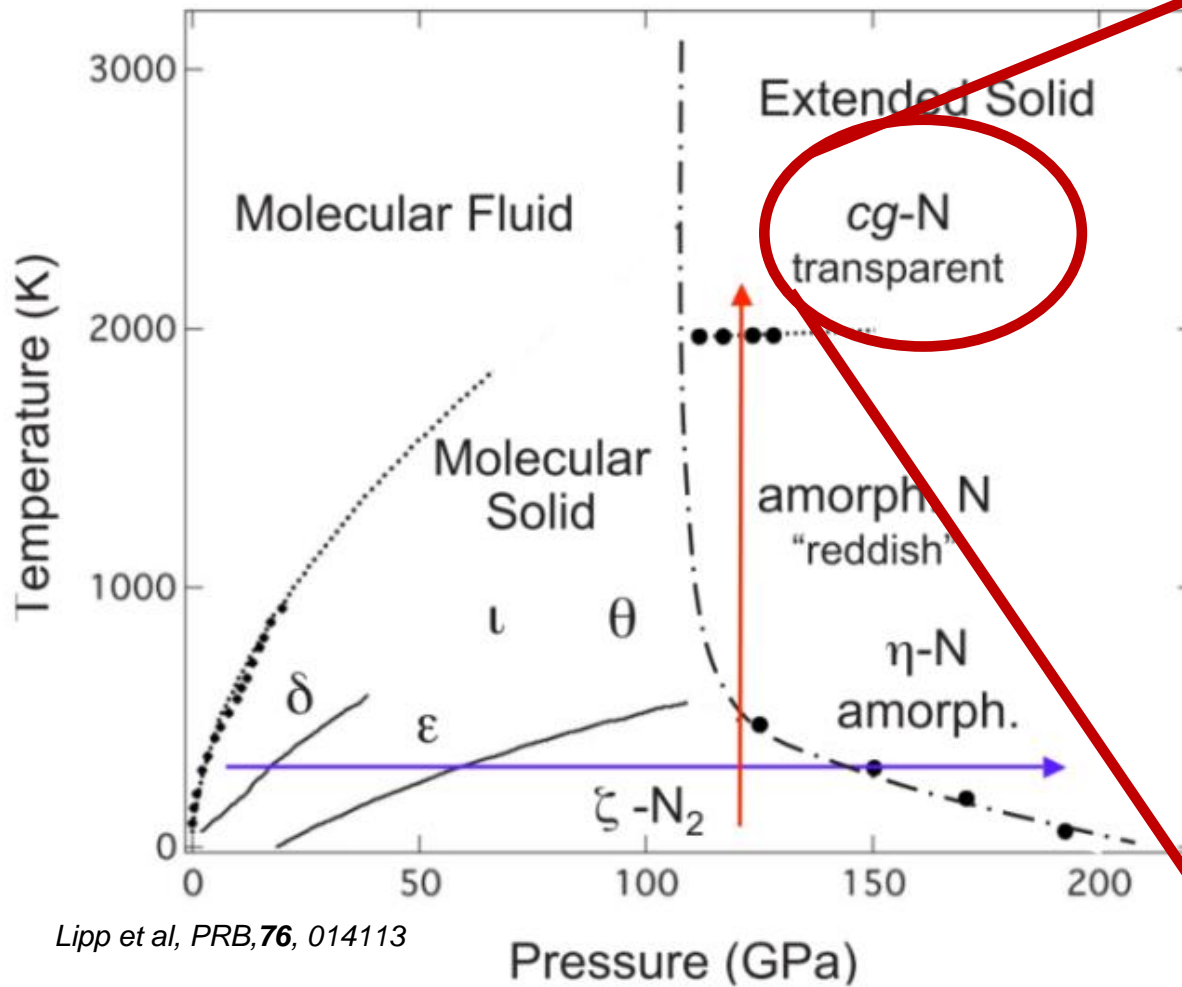
Evolution under pressure



Amorphisation of N₂ at ambient temperature
(Goncharov et al, 2000)

➤ Interpretation in terms of a "non-molecular" but amorphous phase

Evolution under pressure

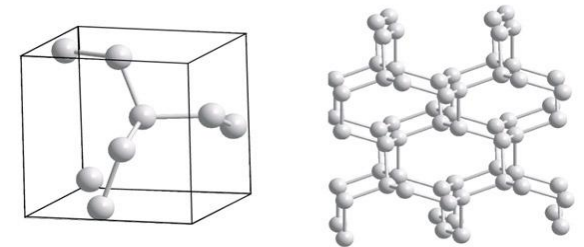


Lipp et al, PRB, 76, 014113

Synthesis of the
crystalline form
(Eremets et al, 2004)

➤ P = 110 GPa et T > 2000 K

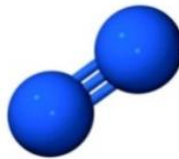
➤ « cg-N » structure



➤ Metastable at ambient
temperature down to
42 GPa

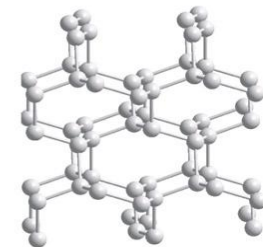
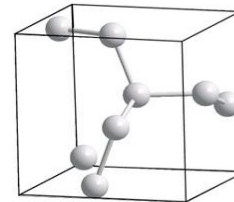
A HIGH ENERGY DENSITY MATERIAL

- **N₂ triple bond : 9.8 eV/atom**



- **N-N (polymeric) single bond : 1.7 eV/atom**

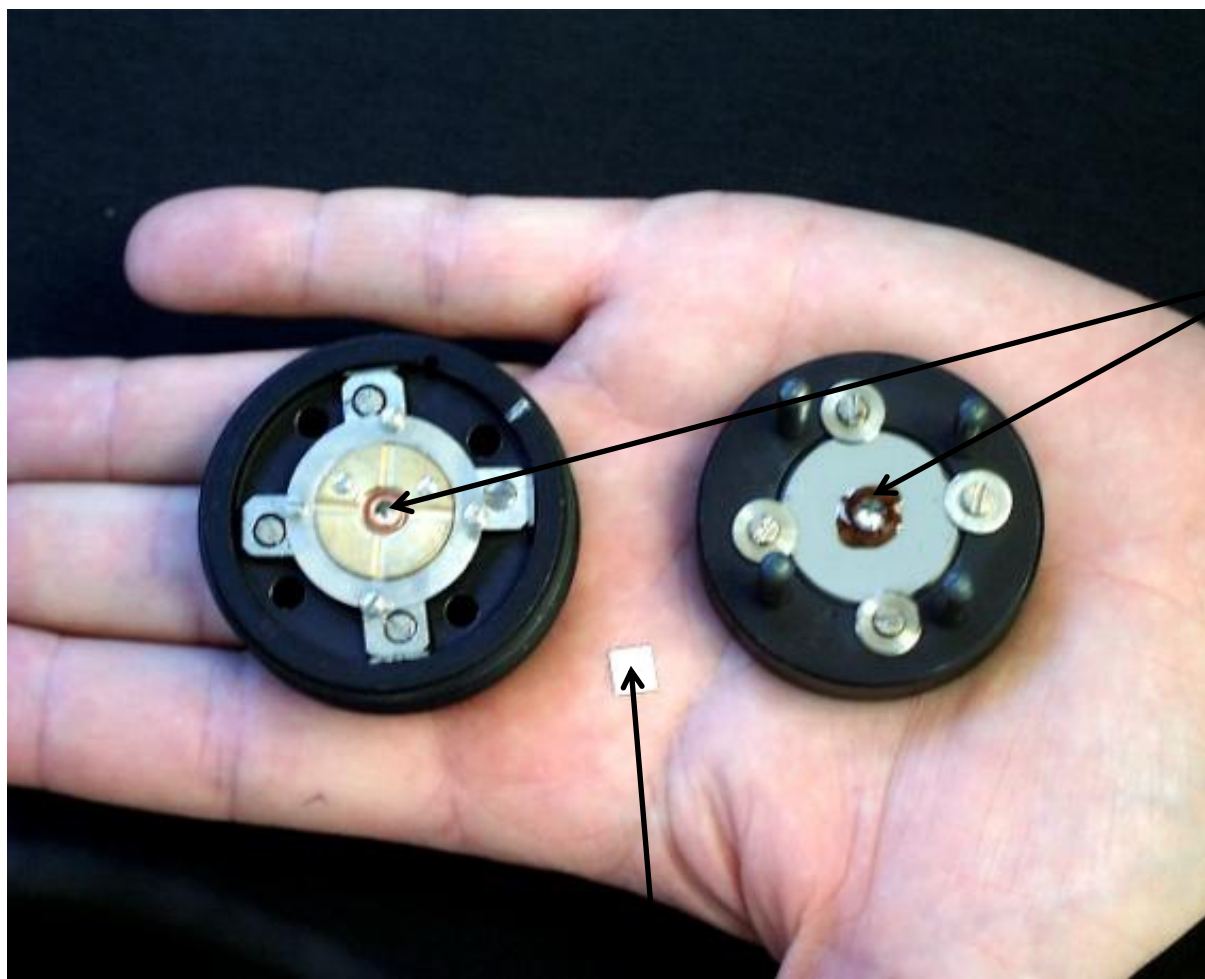
Cg-N



Eremets et al, Nature Mat., 3, 558 (2004)

- **N-polymere : 64 MJ/L**
- **Diesel : 36 MJ/L**
- **TNT : 7.6 MJ/L**
- **O₂-H₂ : 2.7 MJ/L**
- **(Uranium : 1.5 10⁹ MJ/L)**

THE DIAMOND ANVIL CELL



Diamonds

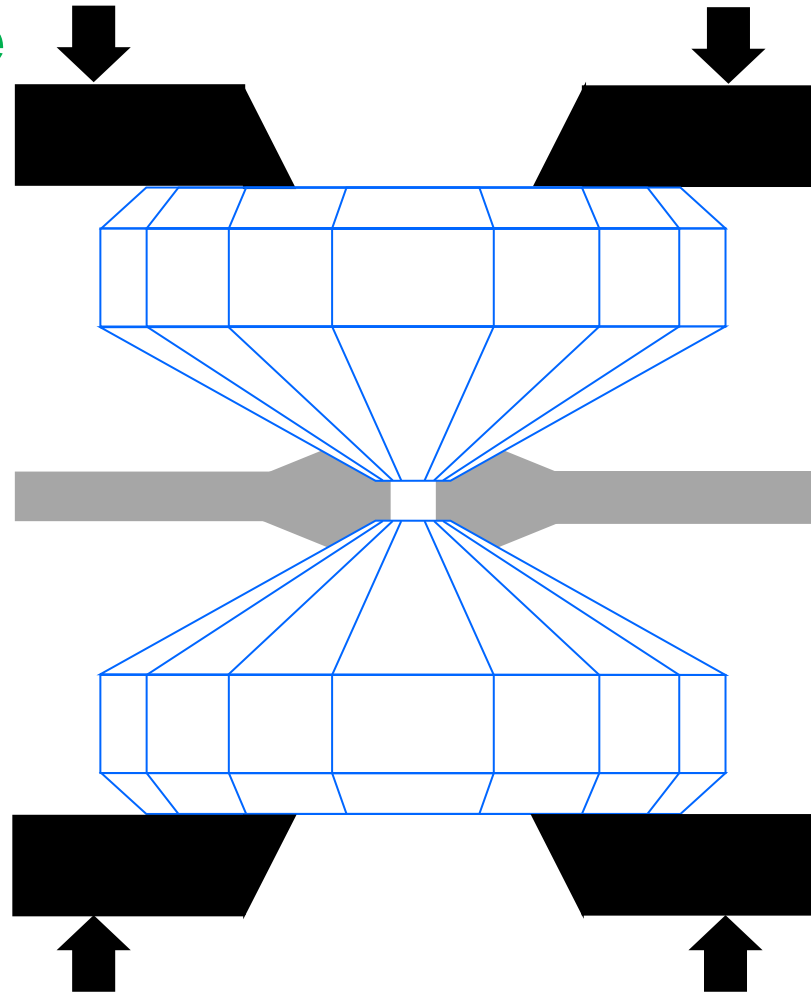
Metallic gasket

THE DIAMOND ANVIL CELL

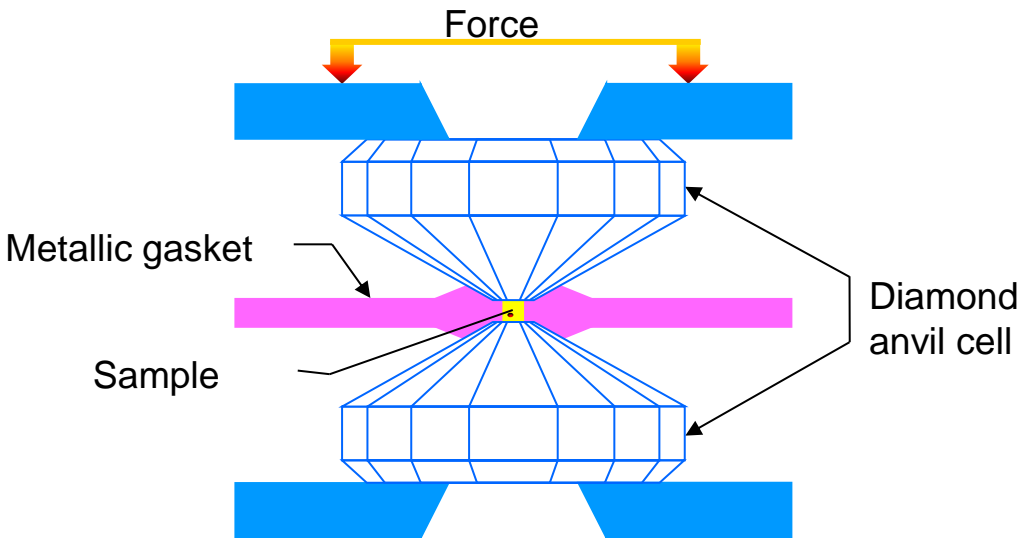
100 bars on the membrane
($S \sim \text{cm}^2$)



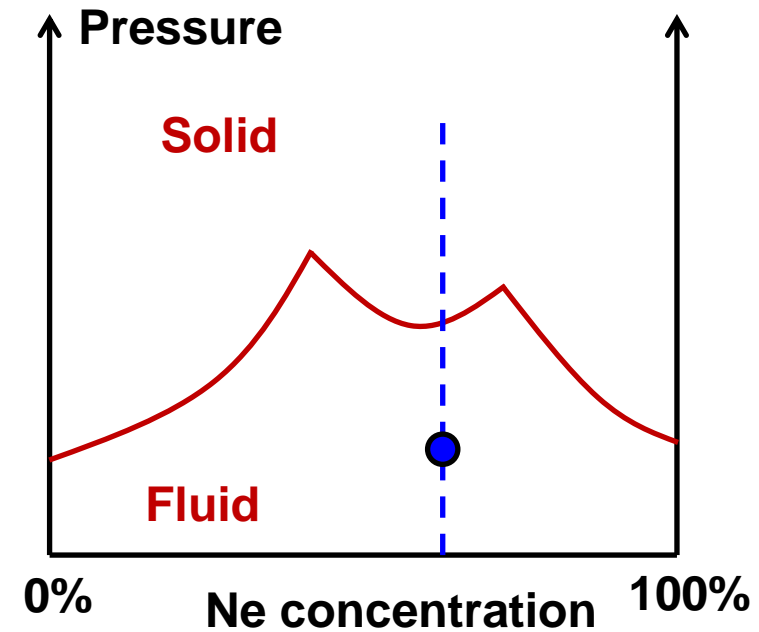
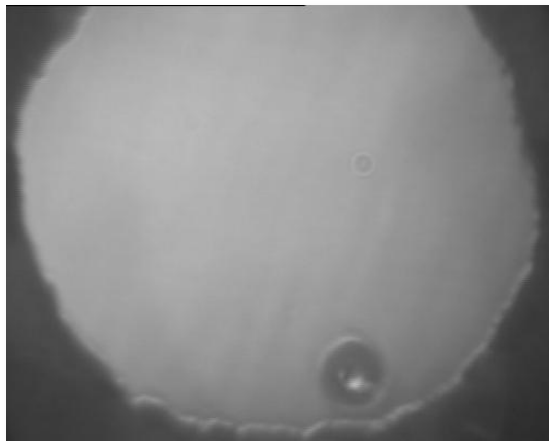
1Mbar in the sample
($S \sim 100\mu\text{m} \times 100\mu\text{m}$)



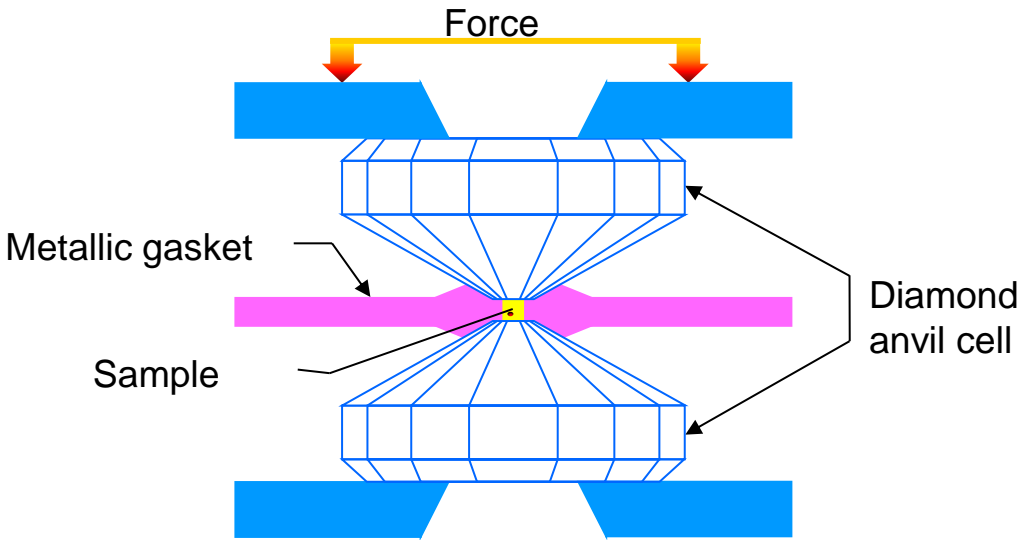
EVOLUTION UNDER PRESSURE



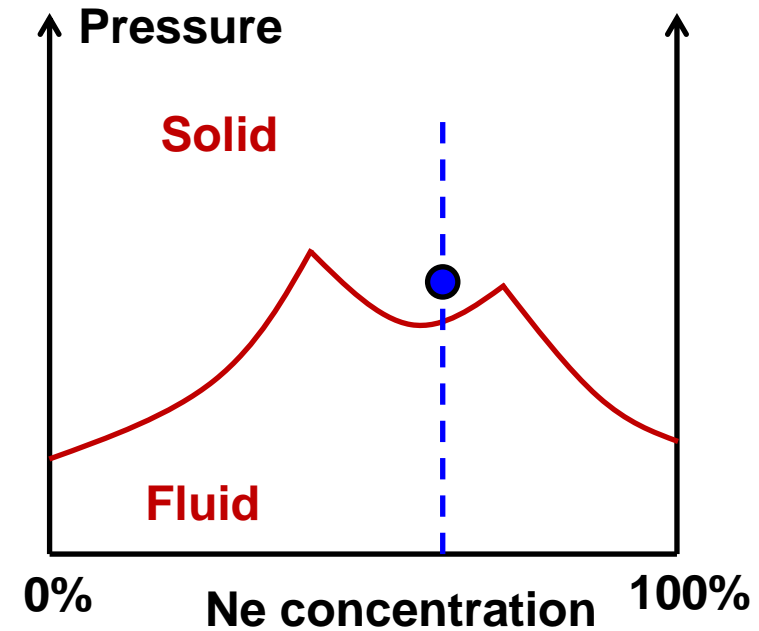
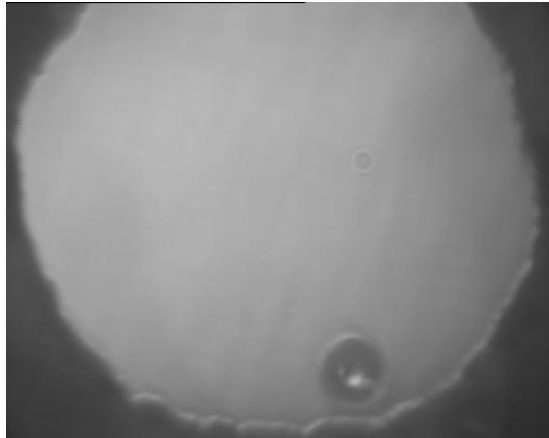
State : homogeneous fluid



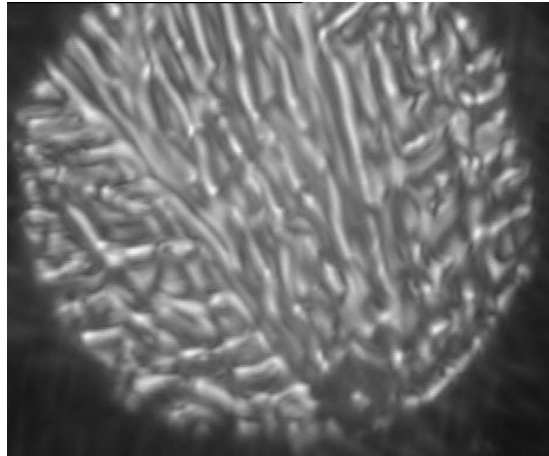
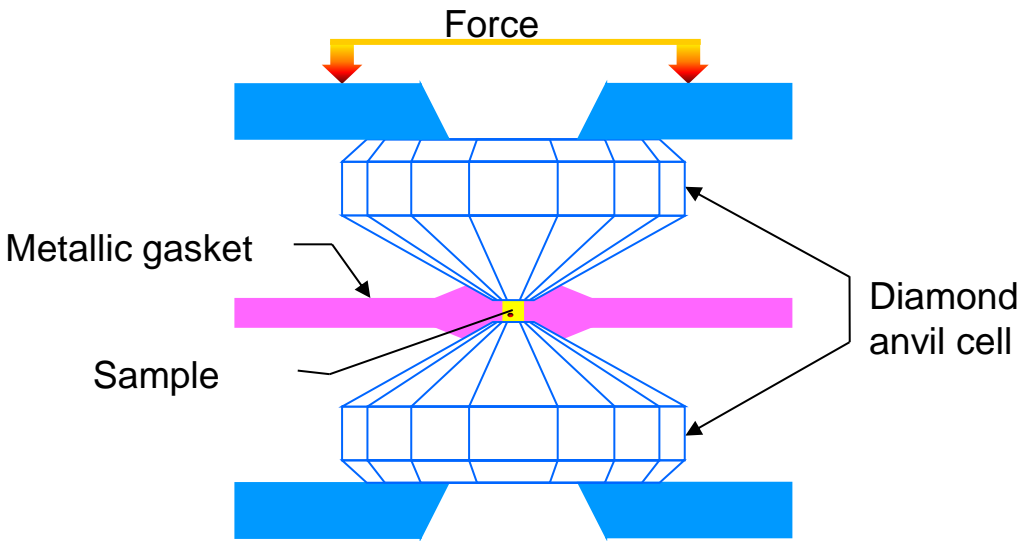
EVOLUTION UNDER PRESSURE



State : Over-pressured fluid

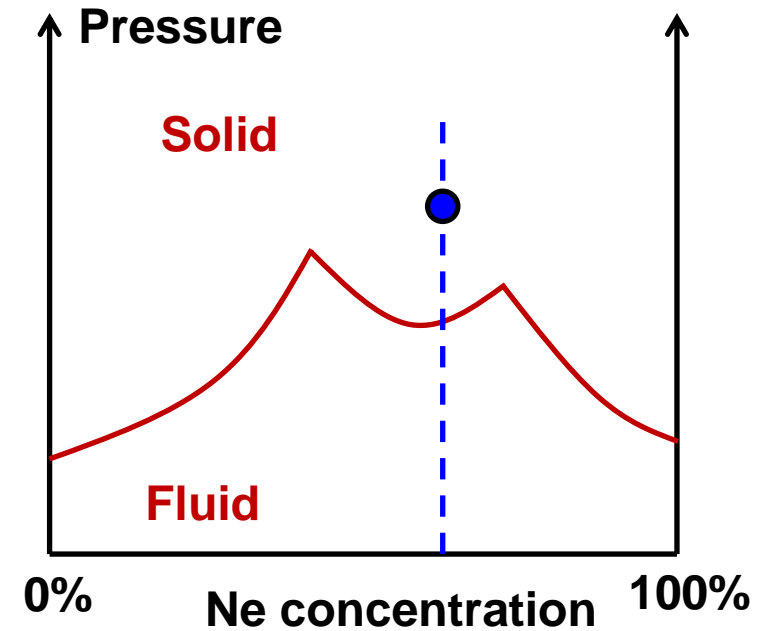


EVOLUTION UNDER PRESSURE

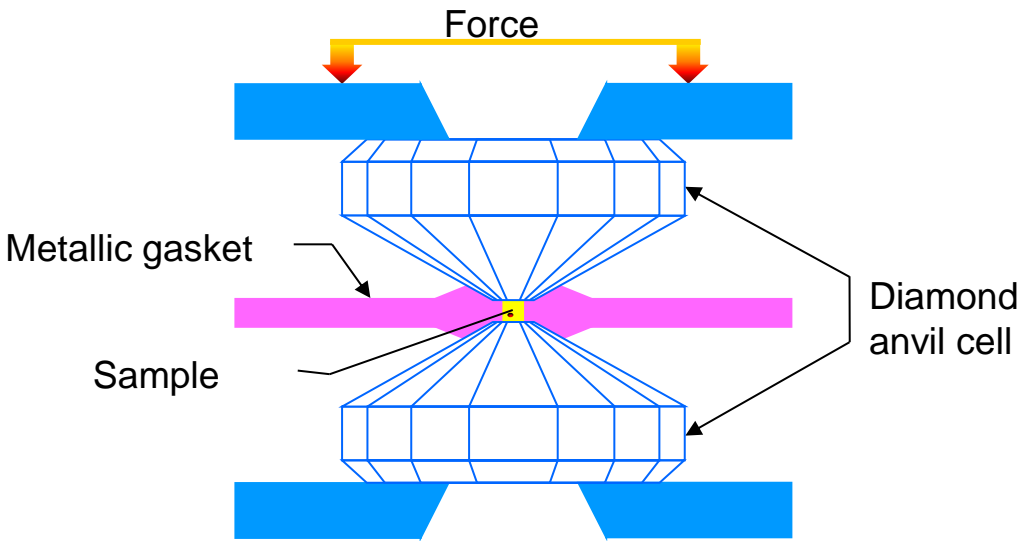


State : Phase separation

Powder or polycrystal

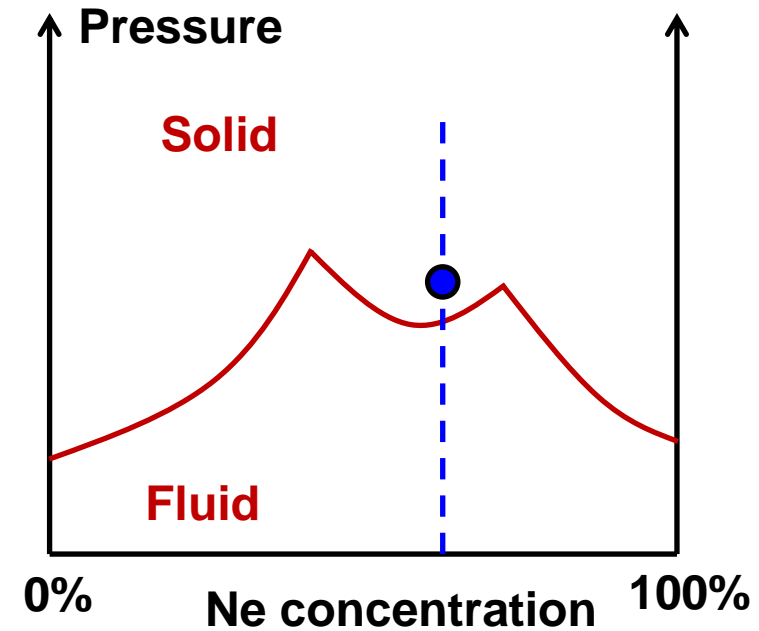
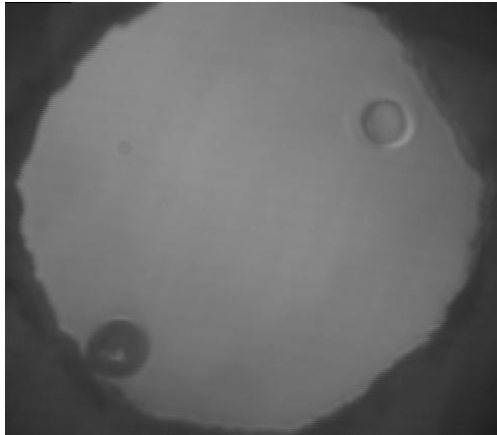


EVOLUTION UNDER PRESSURE

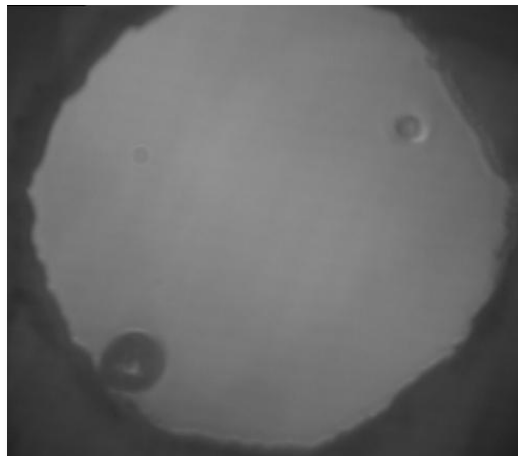
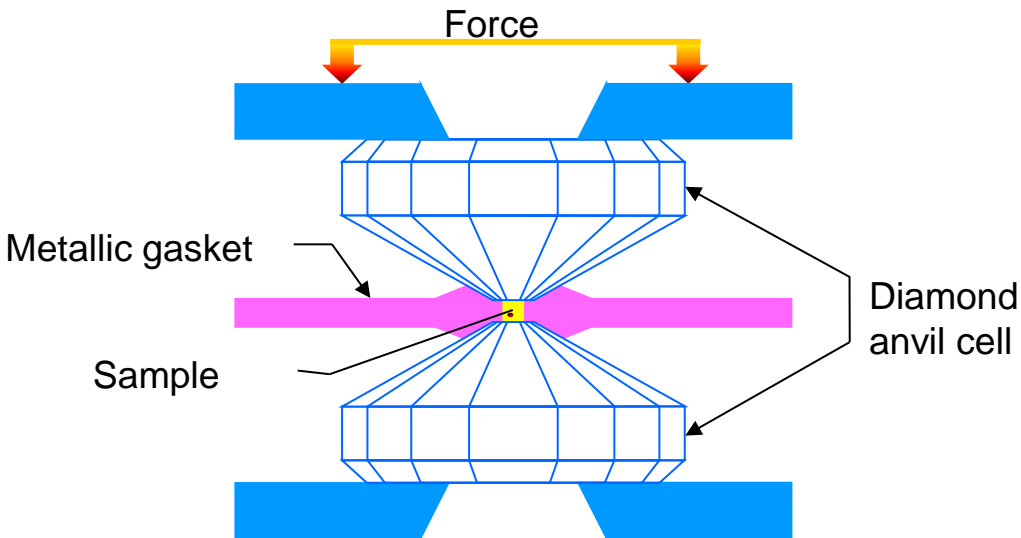


State : Solid-fluid equilibrium

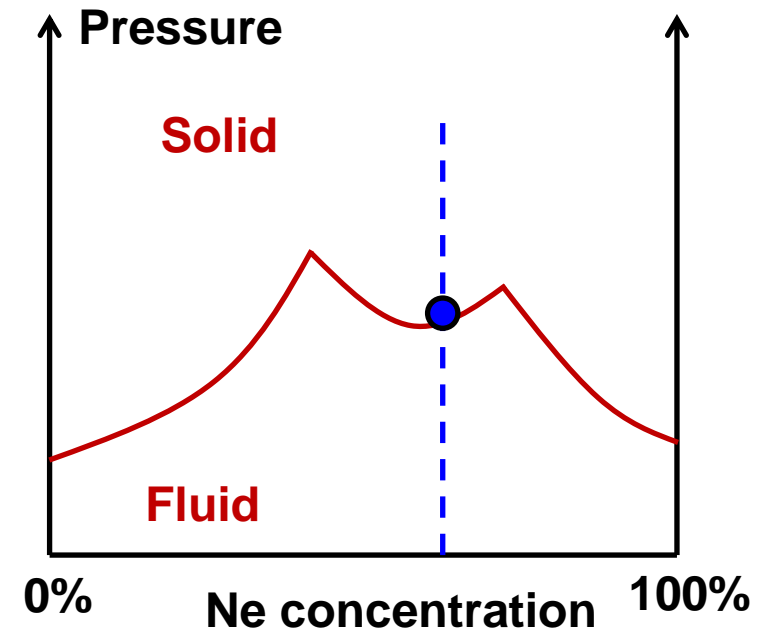
Single - crystal



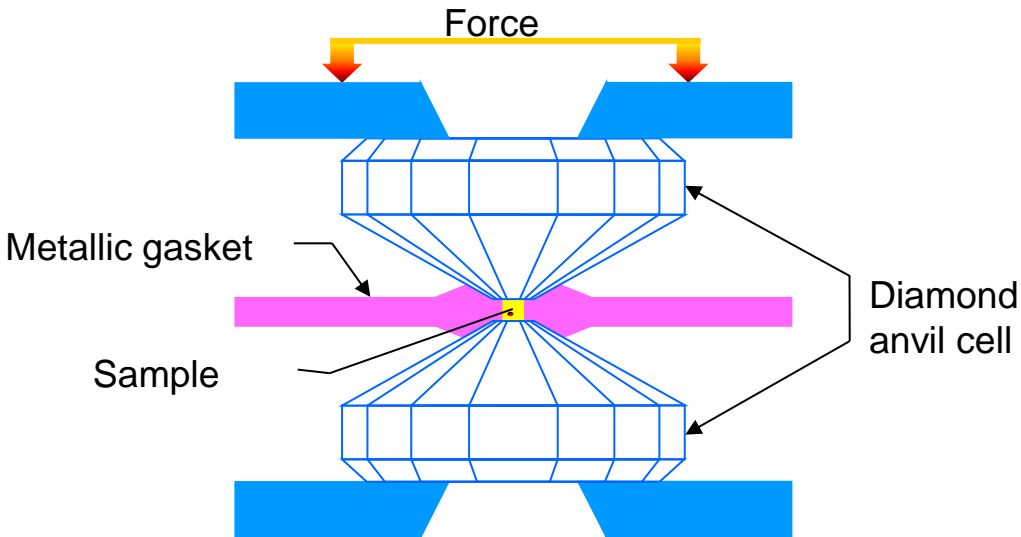
EVOLUTION UNDER PRESSURE



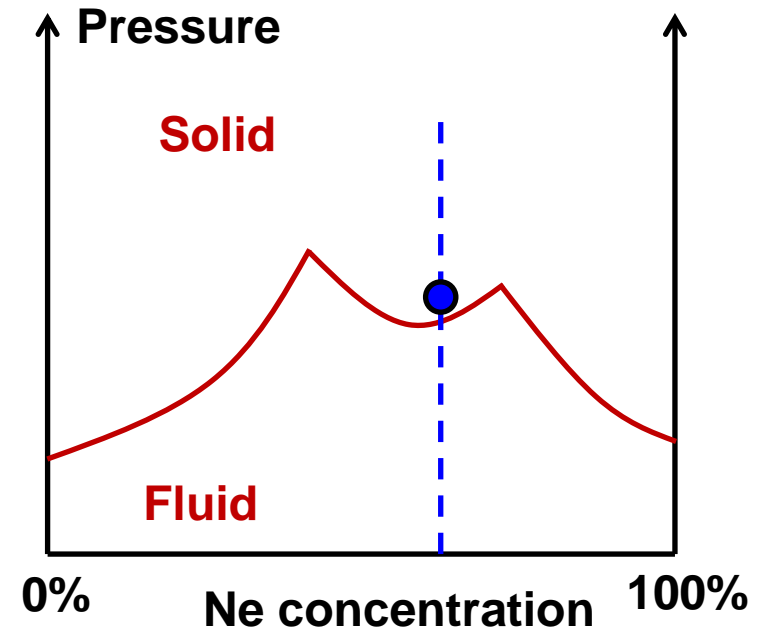
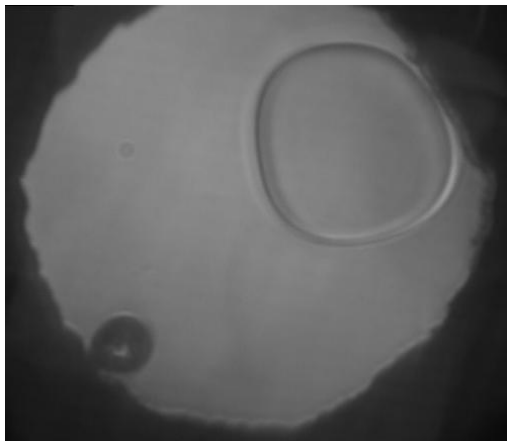
**State : Solid – fluid equilibrium
Threshold of disappearance of
the single-crystal = Liquidus**



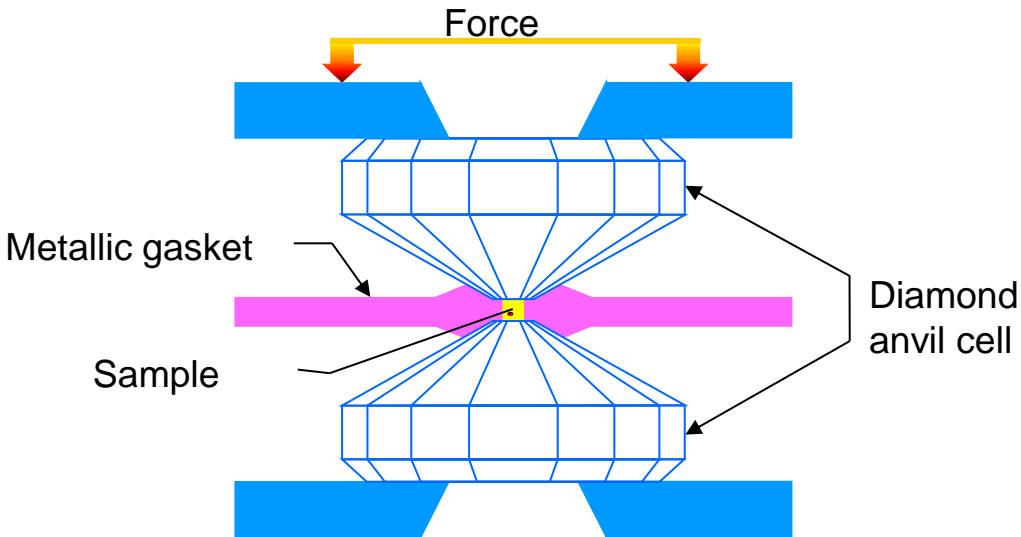
EVOLUTION UNDER PRESSURE



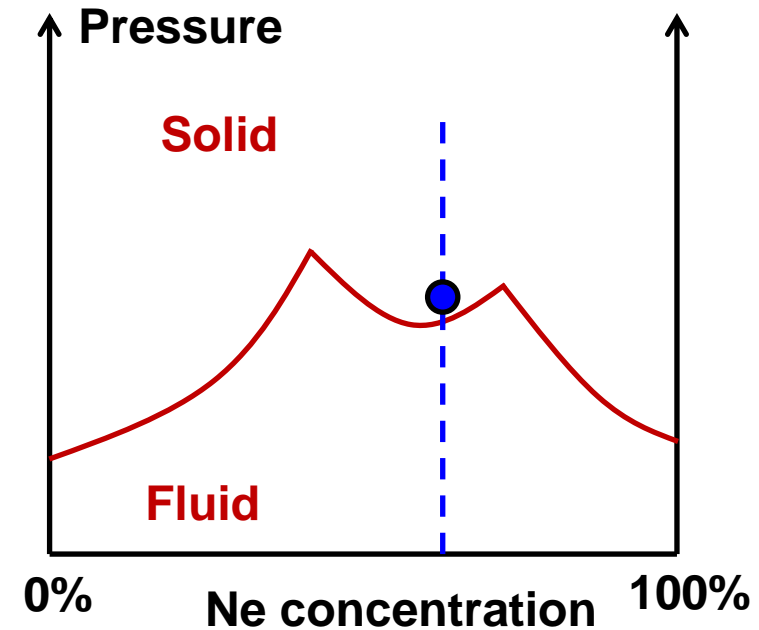
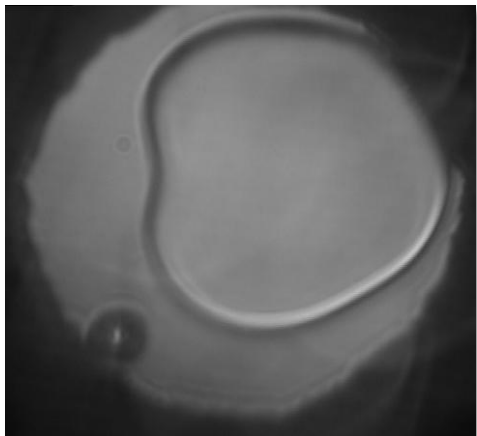
**State : Solid – fluid equilibrium
Single-crystal growth**



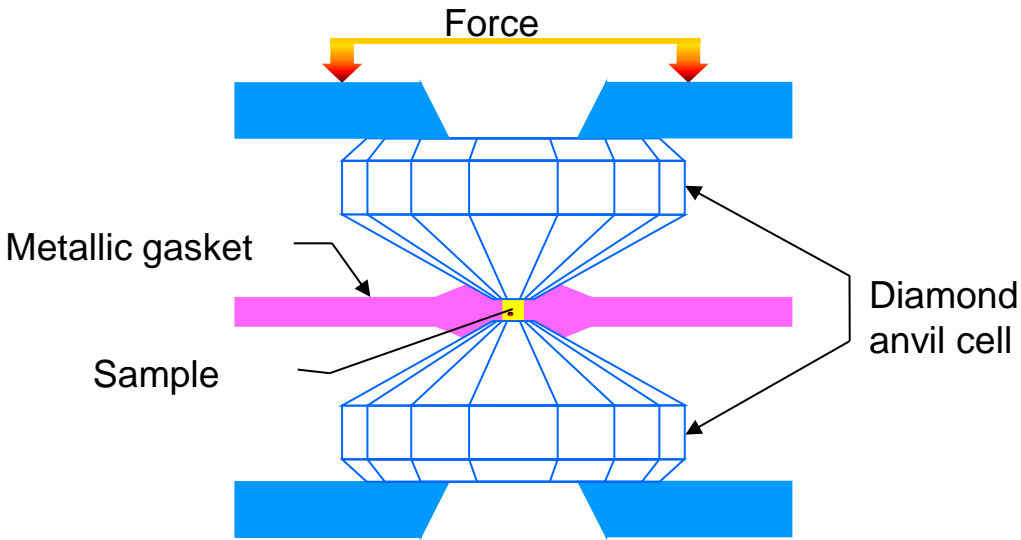
EVOLUTION UNDER PRESSURE



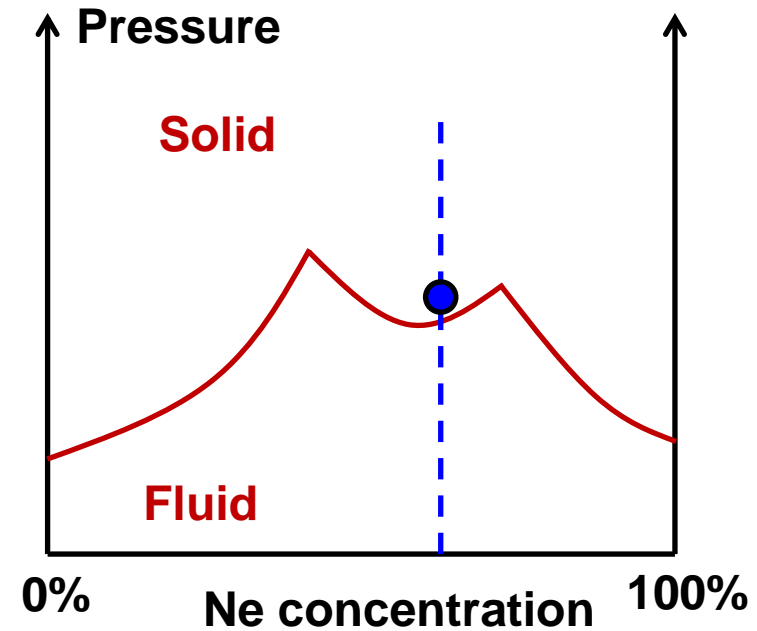
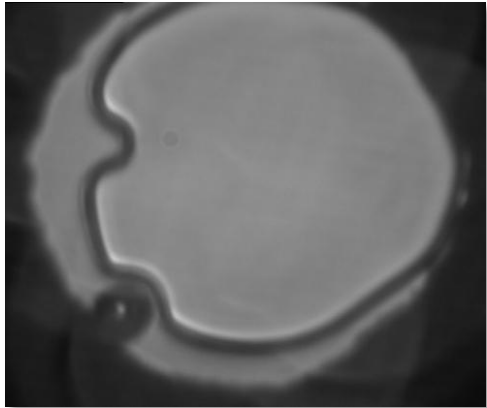
**State : Solid – fluid equilibrium
Single-crystal growth**



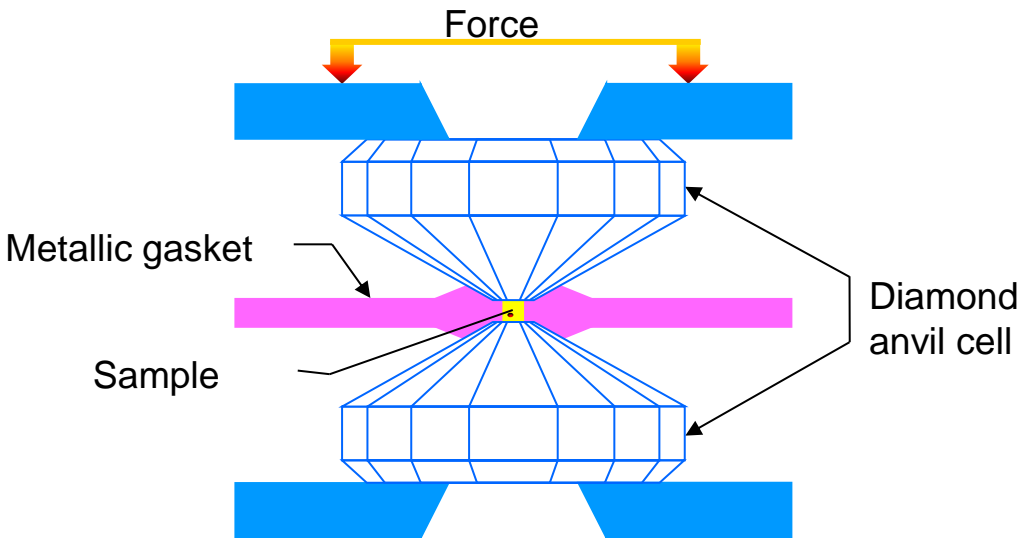
EVOLUTION UNDER PRESSURE



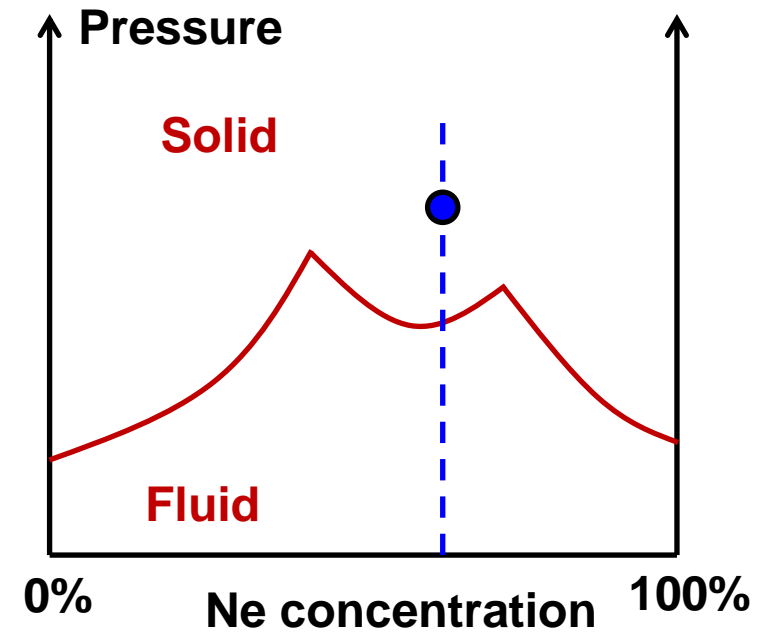
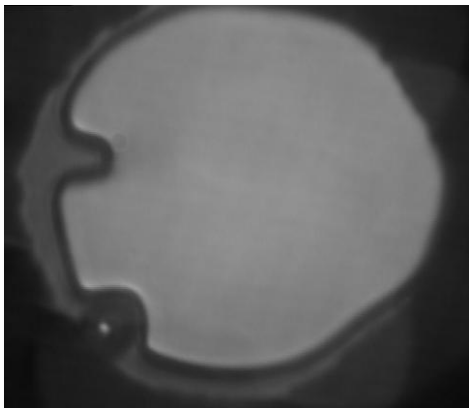
**State : Solid – fluid equilibrium
Single-crystal growth**

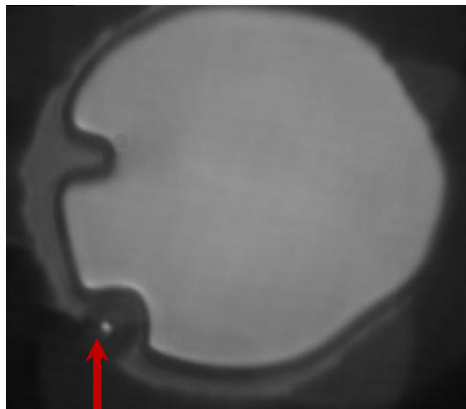


EVOLUTION UNDER PRESSURE



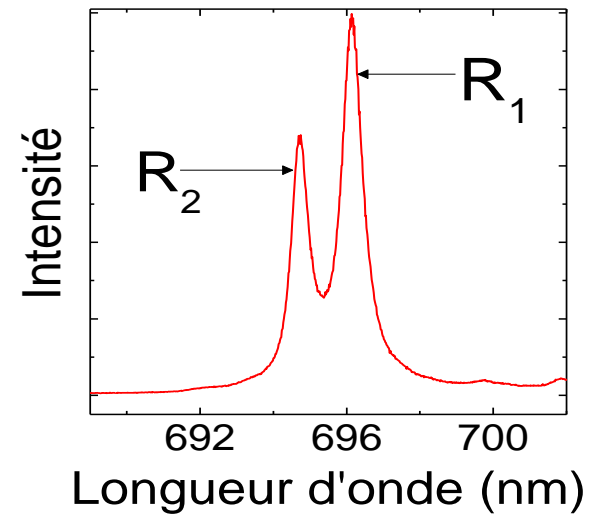
State : Phase separation



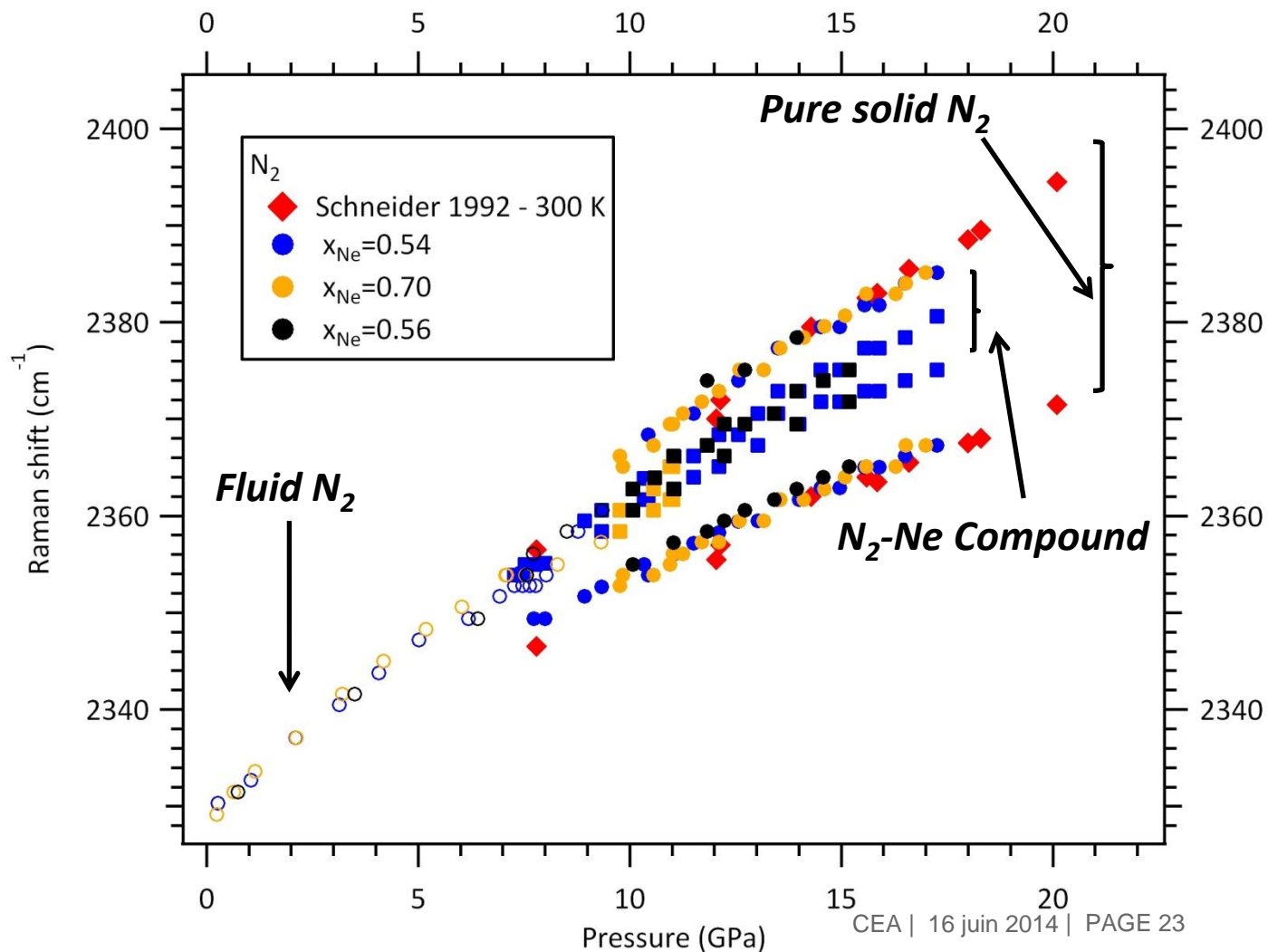
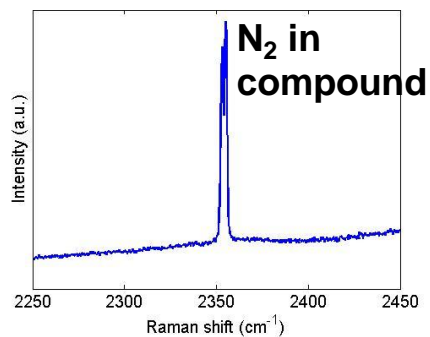
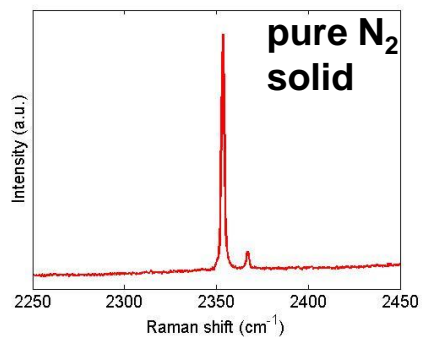
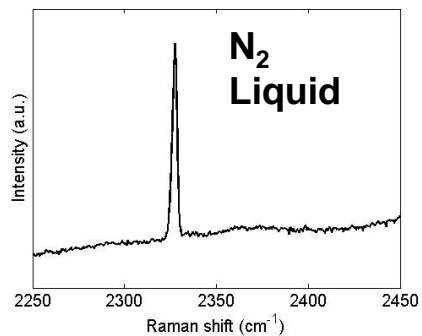


Ruby

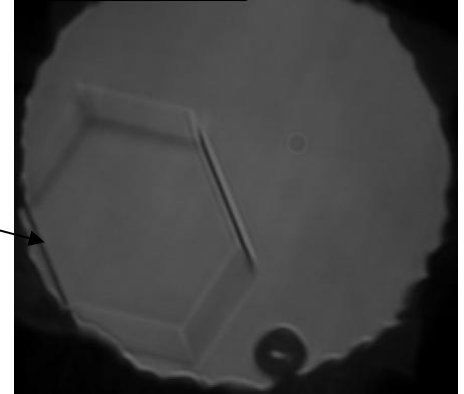
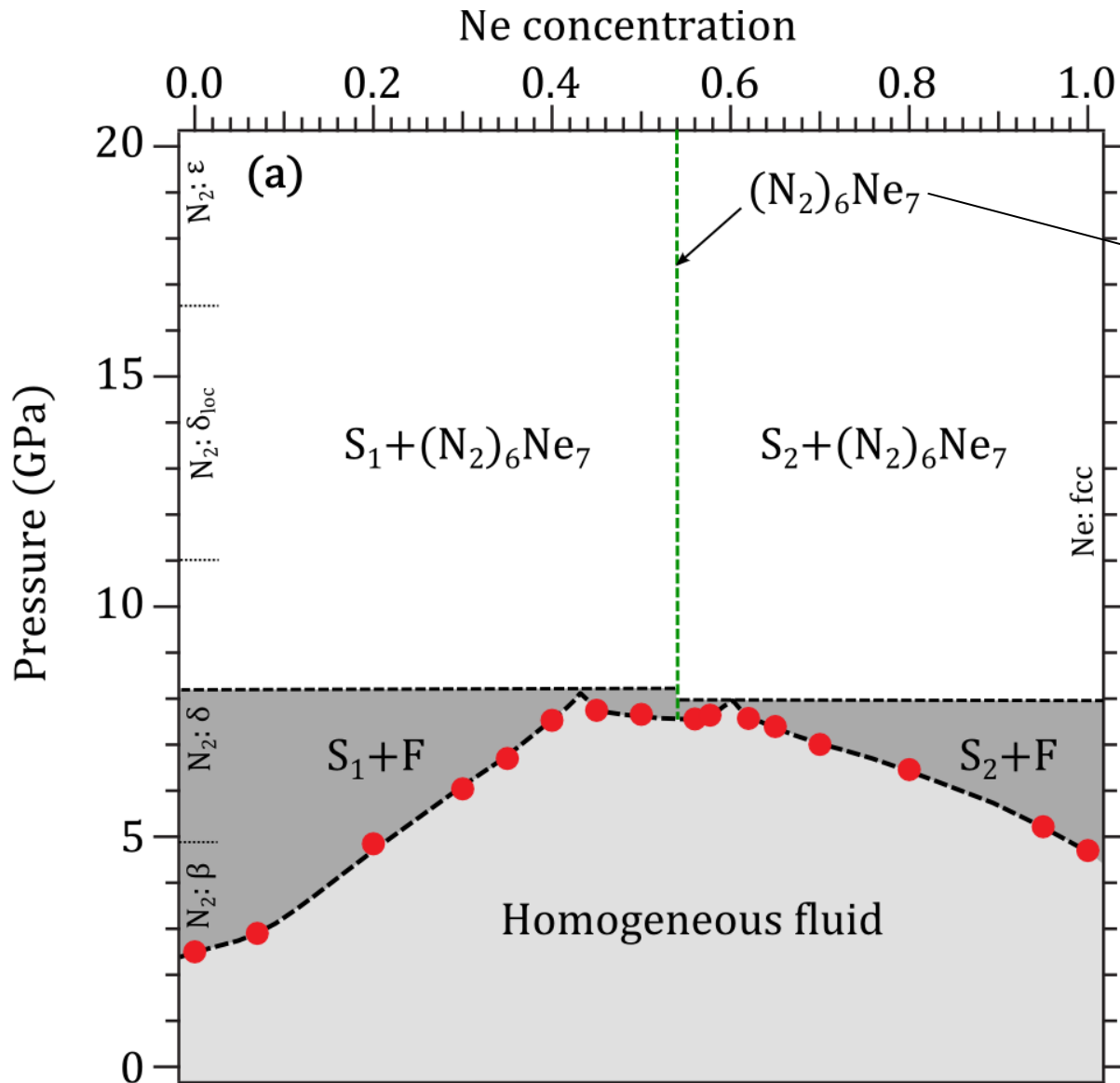
Ruby fluorescence spectrum



Raman spectroscopy : vibrational modes of the N_2 molecule

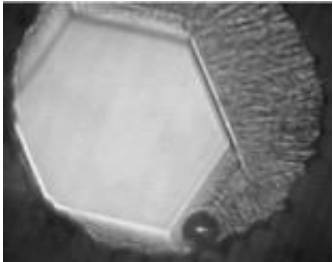


N₂ – NE BINARY DIAGRAM



STRUCTURE OF THE COMPOUND

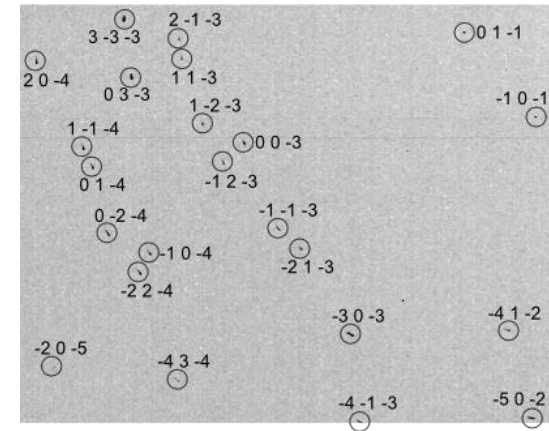
N_2 -Ne
Single crystal



Synchrotron (ESRF, Grenoble)

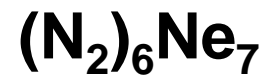
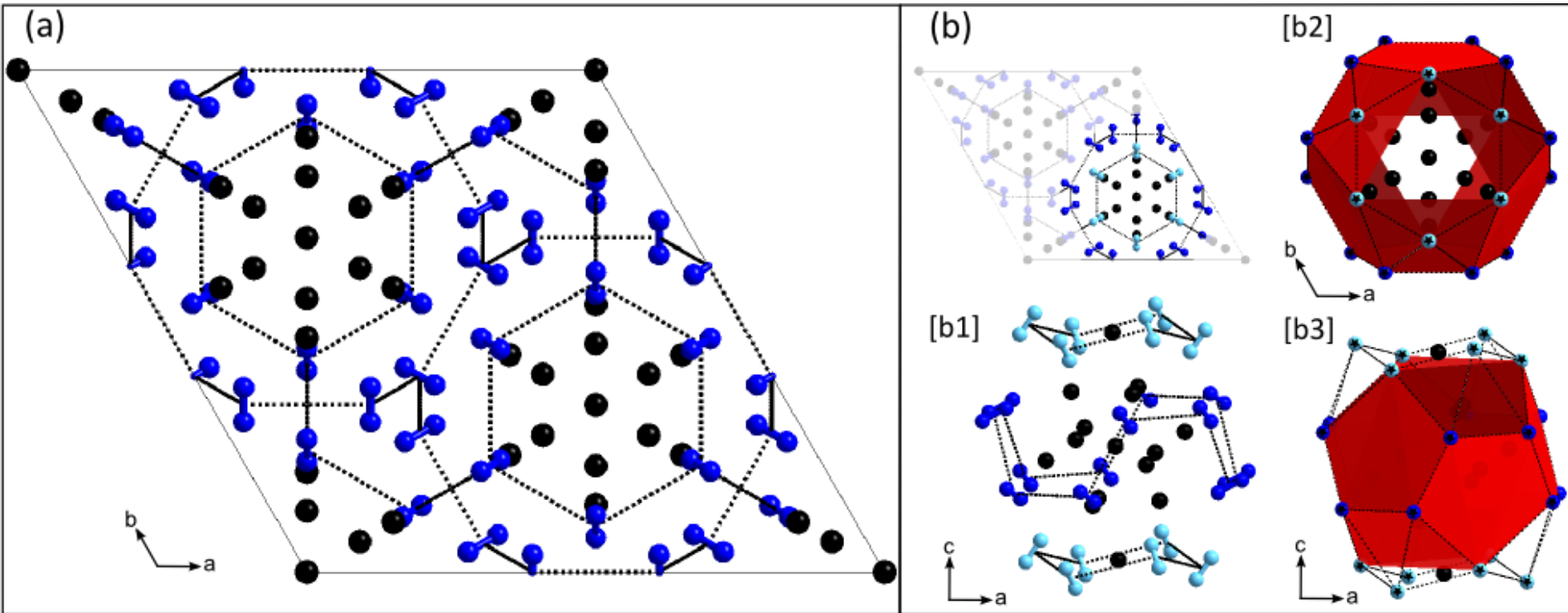


Image plate



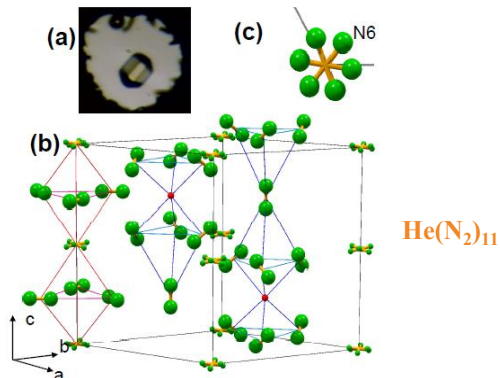
+ Direct methods in crystallography

STRUCTURE OF THE COMPOUND



Hexagonal / $R\bar{3}m$
 $a=b=14.4 \text{ \AA}$, $c=8.09 \text{ \AA}$ @ 8 GPa

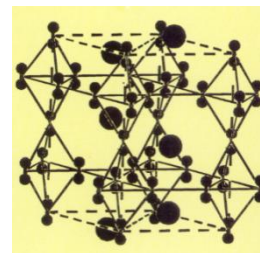
AN ORIGINAL VAN DER WAALS COMPOUND



Structures of (N₂)₁₁He compound to 175 GPa: a close relationship with pure N₂

S. Ninet,^{1,2} G. Weck,² P. Loubeyre,² and F. Datchi¹

¹CEA, DAM, DIF, F-91297 Arpajon, France

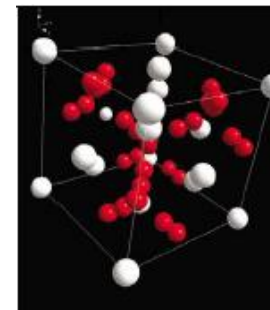


Ar(H₂)₂

VOLUME 72, NUMBER 9 PHYSICAL REVIEW LETTERS 28 FEBRUARY 15

Compression of Ar(H₂)₂ up to 175 GPa: A New Path for the Dissociation of Molecular Hydrogen

Paul Loubeyre, René Letoullec, and Jean-Pierre Pinceaux
Physique des Milieux Condensés, Université Paris 6, boîte 77, 4 place Jussieu, 75252 Paris, France



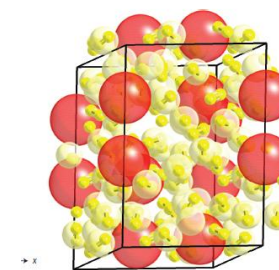
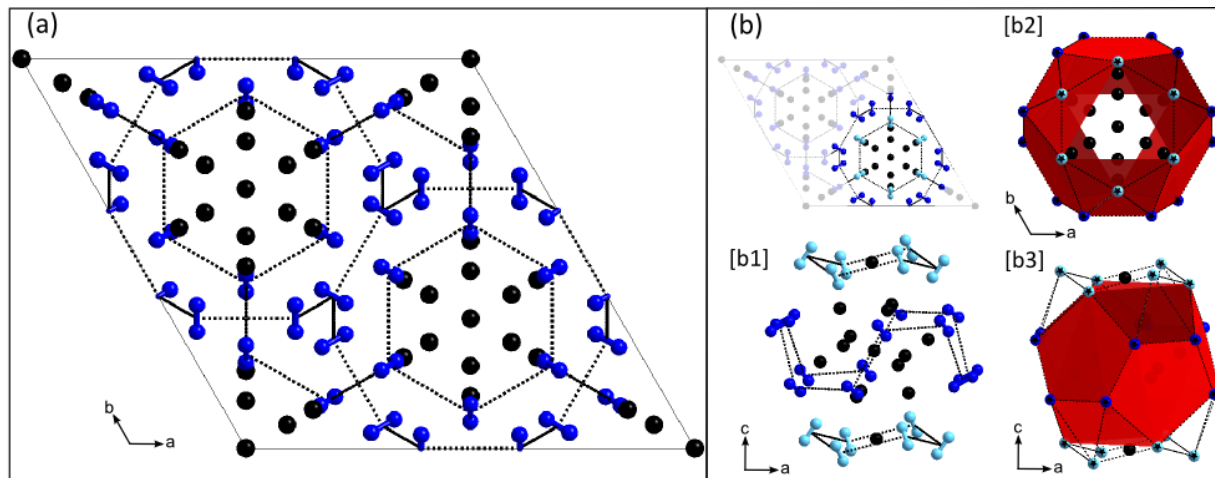
Xe(O₂)₂

PHYSICAL REVIEW B 82, 014112 (2010)

Oxygen/noble gas binary phase diagrams at 296 K and high pressures

Gunnar Weck, Agnès Dewaele, and Paul Loubeyre
 CEA, DAM, DIF, F-91297 Arpajon, France
 (Received 17 May 2010; published 28 July 2010)

The known phase diagrams of O₂ (as well as N₂, Ar, and Xe) mixtures have been measured at 296 K



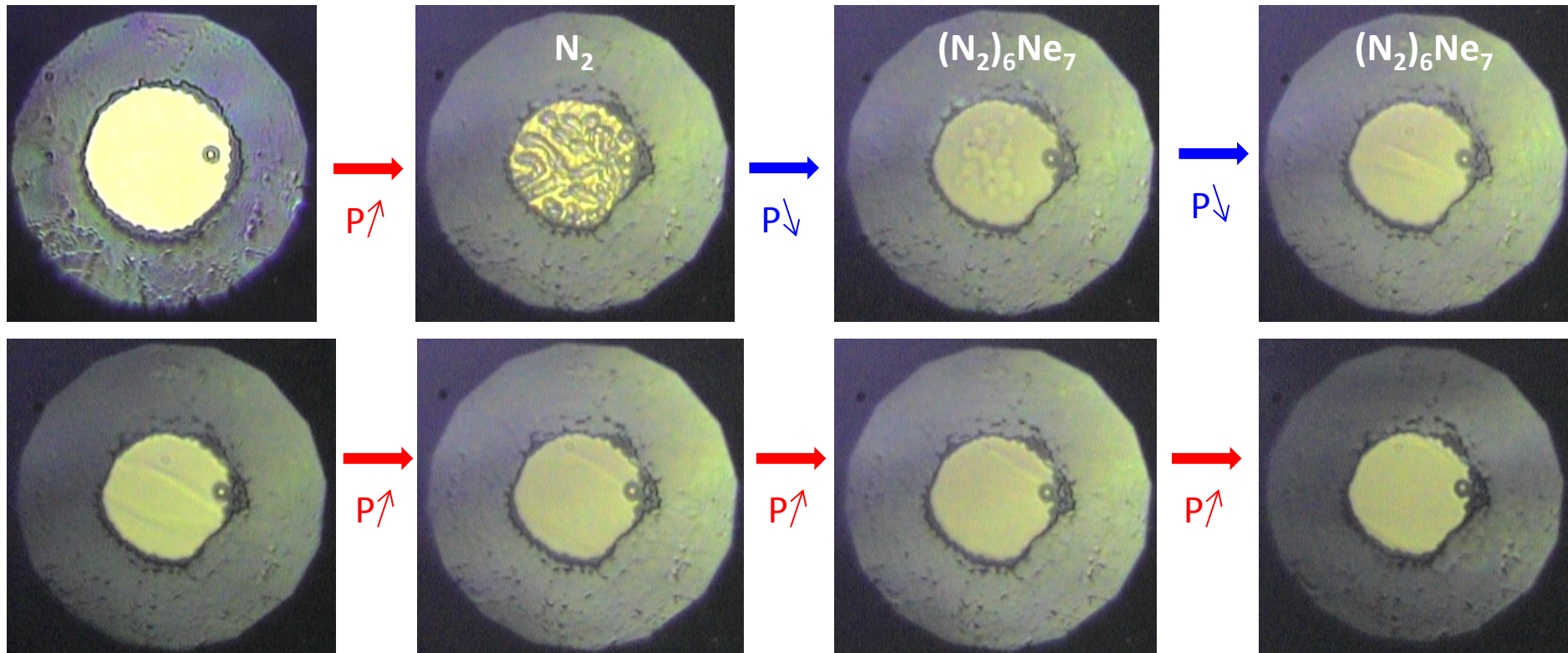
3d lattice structure of Xe(H₂)₇. The xenon atoms are surrounded by

Xe(H₂)₇

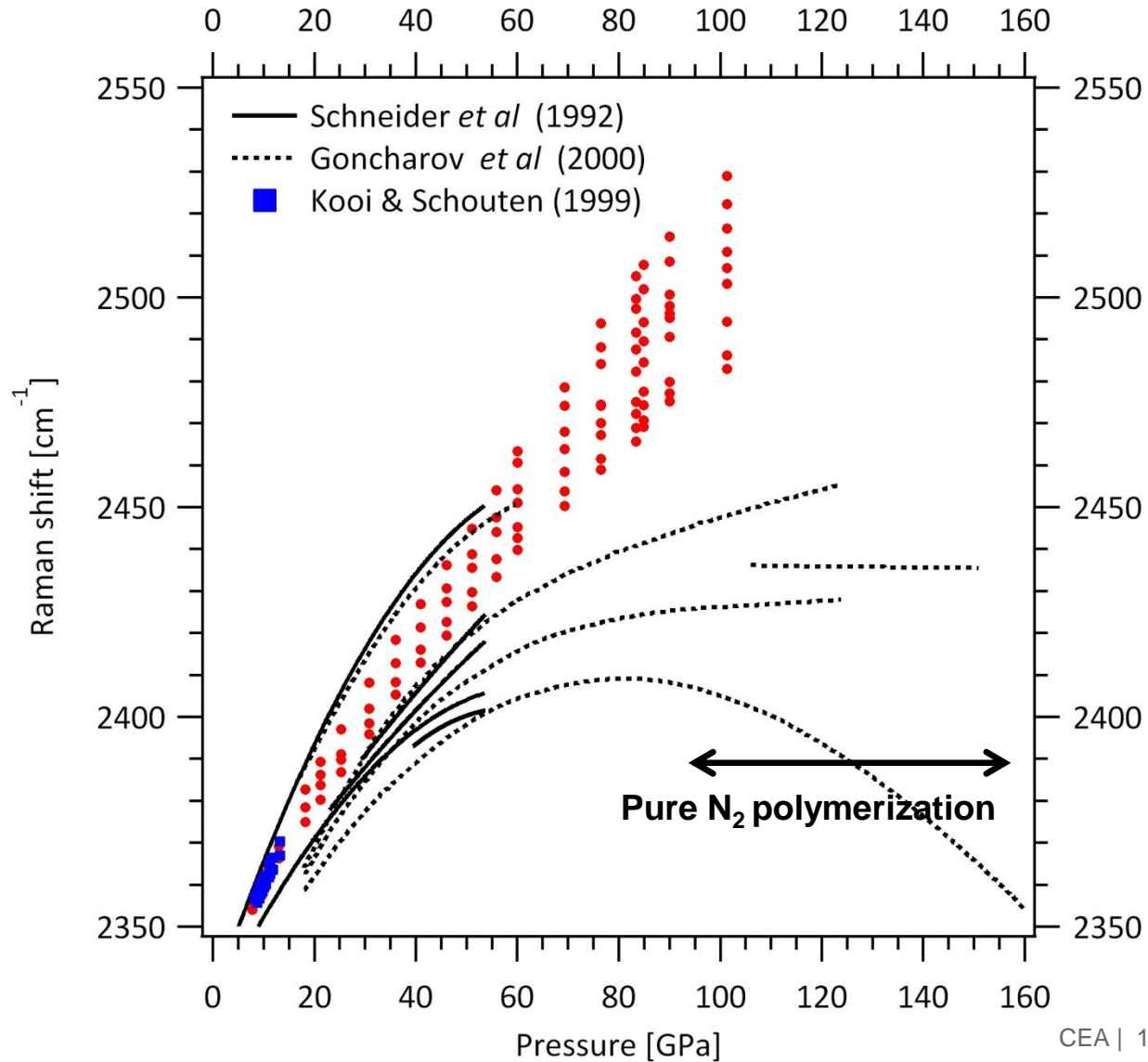
Pressure-induced bonding and compound formation in xenon-hydrogen solids

Maddury Somayazulu^{1*}, Przemyslaw Dera², Alexander F. Goncharov¹, Stephen A. Gramsch¹, Peter Liermann³, Wenge Yang³, Zhenxian Liu¹, Ho-kwang Mao^{2,3} and Russell J. Hemley¹

- Diamonds with 100 μm diameter
- Raman characterization

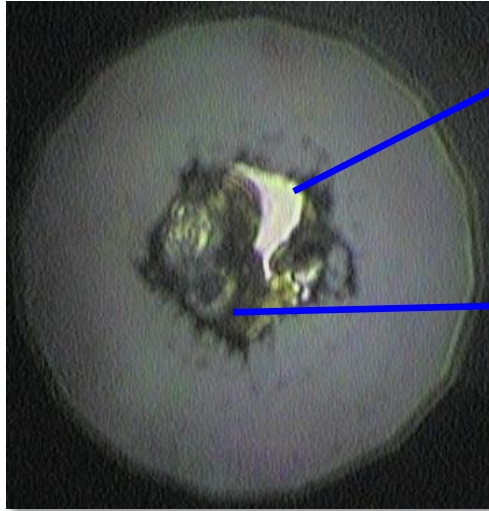


HIGH PRESSURE BEHAVIOR

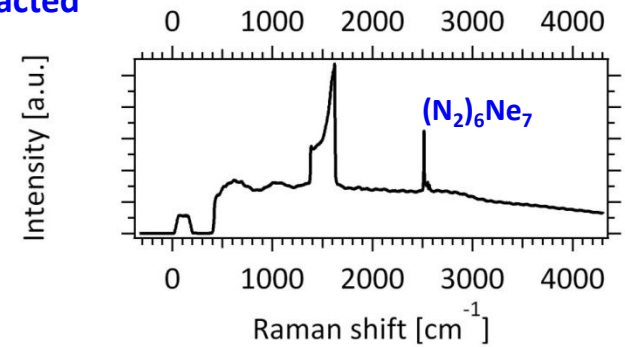


HIGH PRESSURE BEHAVIOR

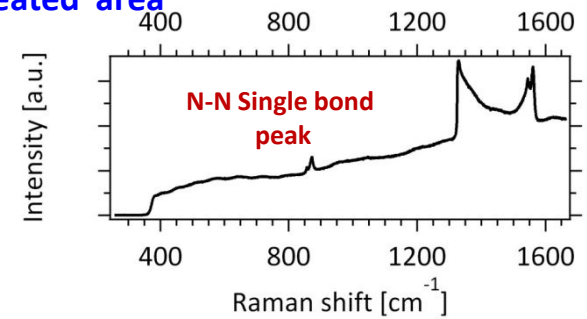
Laser heating at 130 GPa (YAG)



Non-reacted area

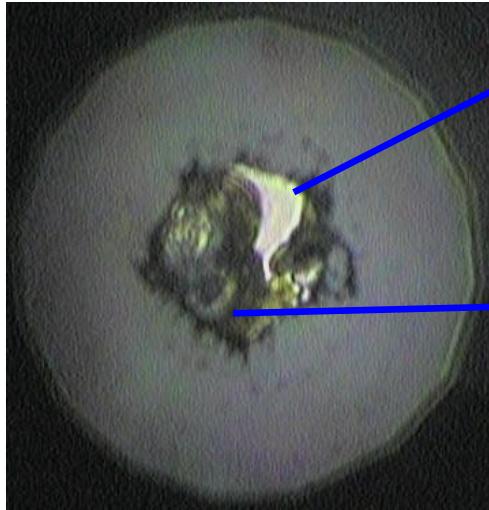


Strongly heated area

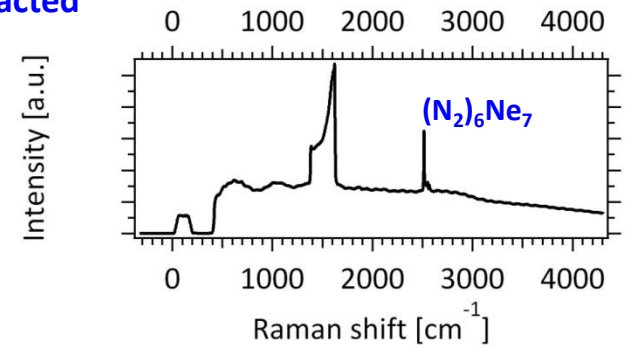


HIGH PRESSURE BEHAVIOR

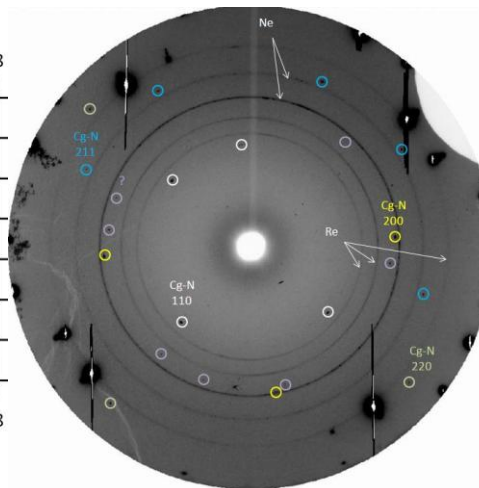
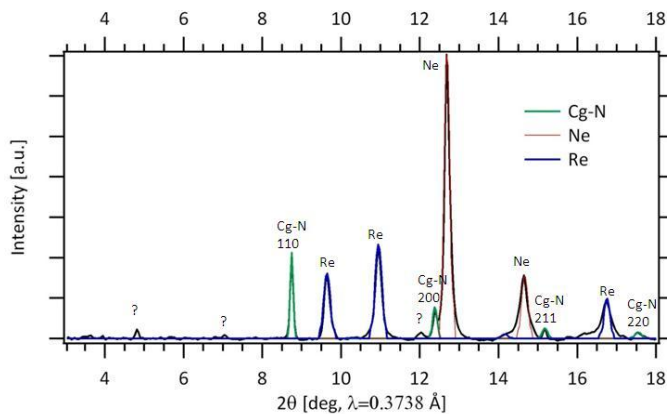
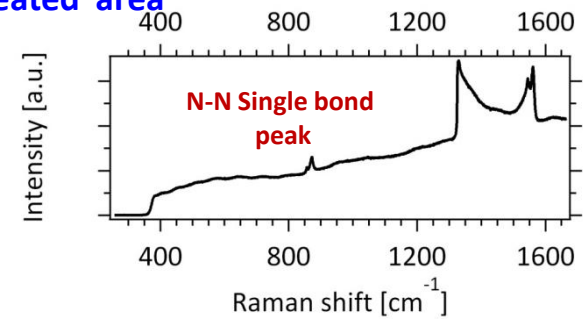
Laser heating at 130 GPa (YAG)



Non-reacted area



Strongly heated area



Phase separation under heating

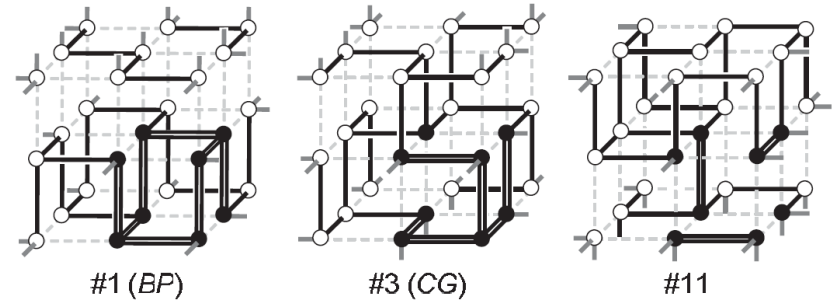
Recrystallisation of cg-N
Neon stays as a powder

New perspectives opened by numerical simulations

- Numerical simulations predict a lower polymerization pressure (65 GPa) than the experimental one (110 GPa)

Pickard et al, PRL 102, 125702 (2009)

Erba et al, PRB 84, 012101 (2011)



Zahariev et al

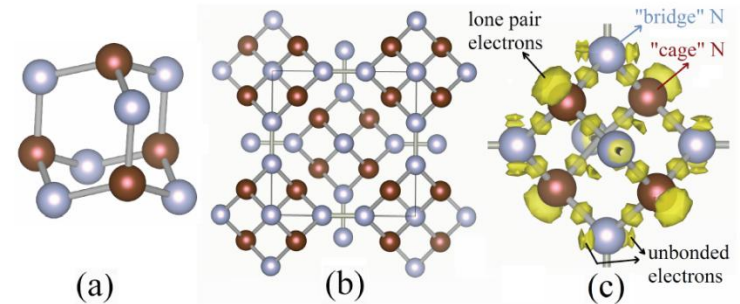
- Atomic phases of nitrogen are distortion of simple cubic ; different structures predicted

Zahariev et al, PRL, 97 155503 (2006)

Pickard et al, PRL 102, 125702 (2009)

- Exotic phases predicted : diamondoid N₁₀

Wang et al, PRL, 109, 175502 (2012)



Wang et al

Binary mixtures will provide geometrical constraints and / or nano-structuration with phase separation